


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In disinfection, gaseous chlorine (Cl<sub>2</sub>) or liquid sodium hypochlorite (bleach, NaOCl) is added to, and reacts with, water to form hypochlorous acid. In the presence of bromine, hypobromous acid is also formed. Both chlorine and bromine are in the "halogen" group of elements, and have similar chemical characteristics.

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How to shock or disinfect a well: this article explains how to shock a well, when, why, and exactly how to chlorinate a drinking water well. We provide a table explaining the quantity of bleach needed to disinfect a well, and a table comparing the 3 Common Well Water Disinfectants: Chloramine, Chlorine and Chlorine Dioxide. We include a detailed description of the well shocking procedure using ...

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Swimming Pool Age & Spa Merchandiser, November 1985 PPM or ORP: Which Should Be Used? Water treatment experts are becoming increasingly aware that

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The Niagara Falls, New York, water treatment plant first used chlorine dioxide for drinking water treatment in 1944 for phenol destruction. Chlorine dioxide was introduced as a drinking water disinfectant on a large scale in 1956, when Brussels, Belgium, changed from chlorine to chlorine dioxide. Its most common use in water treatment is as a pre-oxidant prior to chlorination of ...

## **Chlorine dioxide - Wikipedia**

NEW YORK STATE DESIGN STANDARDS FOR INTERMEDIATE SIZED WASTEWATER TREATMENT SYSTEMS MARCH 5, 2014 . New York State . Department of Environmental Conservation . Division of Water . 625 Broadway . Albany, New York 12233-3505

## **New York State Department of Environmental Conservation**

Chloramine is used as a disinfectant for water because it is less aggressive than chlorine and more stable against light than hypochlorites. Drinking water disinfection.  $\text{NH}_2\text{Cl}$  is commonly used in low concentrations as a secondary disinfectant in municipal water distribution systems as an alternative to chlorination. This application is increasing.

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